# Synthesis and Characterization of Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub>/LaMnO<sub>3</sub> **Composites as Microwave Absorbers**

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Article Info	ABSTRACT
Article history: Received Jul 25, 2021 Revised Sep 25, 2021 Accepted Oct 29, 2021	The synthesis and characterization of Ba <sub>0.6</sub> Sr <sub>0.4</sub> Fe <sub>12</sub> O <sub>19</sub> /LaMnO <sub>3</sub> composite material has been successfully carried out by mechanical alloying method using high energy milling. Crystal structure and surface morphology were characterized using x-ray diffraction and scanning electron microscopy. While the value of magnetization and the ability to absorb microwaves, vibrating cample magnetization and vactor naturate analyzing ware used respectively.
<i>Keywords:</i> Composite hexaferrite-perovskite absorber microwave hard-soft magnetic	Sample magnetization and vector network anaryzing were used, respectively. With variations in weight, does not change the crystal structure. The Ba <sub>0.6</sub> Sr <sub>0.4</sub> Fe <sub>12</sub> O <sub>19</sub> phase has a hexagonal structure and the LaMnO <sub>3</sub> phase has an orthorhombic structure. Surface morphology has a heterogeneous size in the range of 200-450 nm with the form of platelets. The best composite material Ba <sub>0.6</sub> Sr <sub>0.4</sub> Fe <sub>12</sub> O <sub>19</sub> /LaMnO <sub>3</sub> (75/25 wt%) is with composition of 75/25 wt%, and has magnetic properties with a magnetic saturation of Ms ~ 46.83 emu/g, Mr ~ 28.8 emu/g, and a coercive field of Hc ~ 3.88 kOe. The minimum reflection loss value is $-13.0$ dB at 11.2 GHz frequency and broadband absorber with RL of -8dB at 8-12 GHz.

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#### **INTRODUCTION** 1.

Hexaferrite-based materials, especially M-type hexaferrite (MFe<sub>12</sub> $O_{19}$ , M = Ba or Sr) are still a concern of researchers [1-3]. At room temperature type-M hexaferrite material is hard magnetic, has high magnetic saturation Ms, wide Hc coercivity field, strong uniaxial magnetic anisotropy, high curie temperature, and good chemical stability. Therefore, this material is very applicable in technology, for example as a recording medium [1], ceramic color pigments [2], photocatalysts [3], and microwave absorbers at a frequency of 8-12 GHz [4].

A good material as an absorber of electromagnetic waves must have high permeability and permittivity [5]. So the material is in a hard-soft magnetic state. Therefore, to increase the capability of M-type hexaferrite material as a microwave absorber, several methods were carried out, for example combining Ba<sup>2+</sup> and  $Sr^{2+}$  ions (Ba<sub>1-x</sub>Sr<sub>x</sub>Fe<sub>12</sub>O<sub>19</sub>) [6], substituting Mn<sup>2+</sup> ions on Fe<sup>3+</sup> ions (Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12-z</sub>Mn<sub>z</sub>O<sub>19</sub>) [4], making hexaferrite composites with perovskite ( $SrFe_{12}O_{19}-La_{1-x}Sr_xMnO_3$ ) [7],  $BaFe_{12}O_{19}-NiFe_2O_4$  composites [8], or making multiple layers consisting of BaFe<sub>12</sub>O<sub>19</sub>-BaCoZnFe<sub>10</sub>O<sub>19</sub> composites [9].

Several sampling methods that have been used so far are co-precipitation [1, 8], Salt Flux-Assisted [2], auto-combustion [7], mechanical alloying using milling [4,10]. Mechanical alloying method is the easiest and cheapest method.

From the previous study, the substitution of  $Mn^{2+}$  ions on  $Fe^{3+}$  ions ( $Ba_{0.6}Sr_{0.4}Fe_{12}$ -z $MnzO_{19}$ ) with z =0, 1, 2, and 3. They reported that for z = 0 (Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub>) the magnetic properties were Ms = 92.9 emu/g and Hc = 0.424 T, while a maximum reflection loss (RL) of about -4.9 dB in a frequency range of 12.5 GHz

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and RL of -2 dB in a frequency range of 8-10 GHz for a sample thickness of 1.5 mm [4]. In other research for La<sub>1-x</sub>Ba<sub>x</sub>MnO<sub>3</sub> perovskite sample with a sample thickness of 1.5 mm has a value of RL ~ - 2.6 dB at a frequency of 11 GHz [11] with a magnetic saturation value of Ms ~ 9 emu/g and Hc ~ 90 Oe [12]. Meanwhile, the La<sub>1-x</sub>Ca<sub>x</sub>MnO<sub>3</sub> perovskite sample for x = 0.1 has Rl ~ - 42 dB (f = 10.5 GHz) for a sample thickness of 2.0 mm [13]. Based on the results of previous studies, it was stated that Ba/SrFe<sub>12</sub>O<sub>19</sub> material has low reflection loss but the absorption frequency area is relatively wide, while LaMnO<sub>3</sub> material has high reflection loss but relatively narrow absorption. Thus, in this study, a combination of the two materials was carried out to form a composite in the hope of obtaining a material that has a high reflection loss with a relatively wide absorption frequency area. So the purpose of this research is to synthesize and characterize the superior material in the form of Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub>/LaMnO<sub>3</sub> composite as a microwave absorbing material at a frequency of 8-12 GHz using mechanical milling method. This research is an initial study on the development of composite materials which can then be used as a reference for the modification of the two materials so that a more optimal microwave absorbing material can be obtained.

# 2. RESEARCH METHOD

# **2.1. Precursor Preparation**

Raw materials consisting of BaCO<sub>3</sub>, SrCO<sub>3</sub>, and Fe<sub>2</sub>O<sub>3</sub>, each Merck product with purity > 99% were mixed by stoichiometric calculations to form the sample composition  $Ba_{0.6}Sr_{0.4}Fe_{12}O_{19}$ . The equations used are as follows:

 $0.6 \text{ BaCO}_3 + 0.4 \text{ SrCO}_3 + 6 \text{ Fe}_2\text{O}_3 \rightarrow \text{Ba}_{0.6}\text{Sr}_{0.4}\text{Fe}_{12}\text{O}_{19} + \text{CO}_2$  (1) For samples, LaMnO<sub>3</sub> was made from materials La<sub>2</sub>O<sub>3</sub> (Aldrich, 99%) and MnCO<sub>3</sub> (Merck, 99%) with equation (2):

 $La_2O_3 + 2 MnCO_3 \rightarrow 2 LaMnO_{3-\partial} + 2 CO_2$ 

(2)

The Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub>/LaMnO<sub>3- $\partial$ </sub> composite was made with three different compositions, each with a weight % ratio of 85/15, 75/25, and 65/35. Each composite composition was mixed in a stainless steel vial, given iron balls with a ratio of iron:sample weight = 1:1 and added ethanol up to two-thirds. Each sample was milled for 30 hours using high energy milling at 700 rpm with a mechanism of every hour milling, interspersed with 0.5 hours of rest. This is to avoid the heat that occurs due to the impact of the iron balls with the vial. Then the sample was sintered at 1000 °C for 5 hours under atmospheric pressure.

### 2.2. Characterization

X-ray diffractometer (XRD) type PAN Analytical Empyrean ( $\lambda$ Cu-K $\alpha$ , = 1.5406 Å) was used to characterize the phase formation and the composition formed. Data analysis of XRD results both qualitatively and quantitatively using Rietvelt software. The surface morphology was characterized using a scanning electron microscopy (SEM) type-JEOL JED 2300. Meanwhile, the characterization of magnetic magnitude at room temperature and the ability to absorb microwaves were characterized by vibrating-sample magnetometer (VSM) OXFORD VSM 1.2H and Vector Network, respectively. Analysis (VNA) type-Anritsu MS46322A in the 8-12 GHz frequency range.

#### 3. RESULTS AND DISCUSSION

The Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub>/LaMnO<sub>3</sub> composite with a weight ratio of 85/15 wt%, 75/25 wt%, and 65/35 wt% will be referred to as YEG-1, YEG-2, and YEG-3, respectively. Figure 1 shows the results of X-ray diffraction characterization for samples YEG-1, YEG-2, and YEG-3. All samples showed typical peaks of their constituent materials, namely the phases of Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub> (JCPDS card no. 00-051-1879) and LaMnO<sub>3</sub> (JCPDS Card No. 54-1275). See Figure 1(a). Figures 1 (b)-(d) show the results of refining the diffraction pattern using the Rietvelt program. The complete refinement results using the Rietvelt program can be seen in Table 1. In the three samples, the fase  $Ba_{0.6}Sr_{0.4}Fe_{12}O_{19}$  phase has a hexagonal crystal structure, the P63/mmc space group and the LaMnO<sub>3</sub> phase has an orthorhombic crystal structure, the Pbnm space group. The same result was also obtained in our previous study, where  $Ba_{0.6}Sr_{0.4}Fe_{12-3x}Zn_{2x}Ti_xO_{19}$  for x = 0 had a hexagonal crystal structure [14]. Different results were obtained for the LaMnO3 sample, where the crystal structure was hexagonal [15]. According to A. Gholizadeh, LaMnO<sub>3</sub> can have an orthorhombic, rhombohedral or cubic structure, depending on the concentration of Mn<sup>4+</sup> ions [12]. If the Mn<sup>4+</sup> ion content is around 0-12%, it has an orthorhombic structure. In this paper, the Mn ion is in the  $Mn^{2+}$  (MnCO<sub>3</sub>) state which then becomes  $Mn^{3+}$  $(LaMnO_3)$ . Thus the state of the Mn<sup>4+</sup> ion does not exist. From Table 1. it can be seen that as the percentage by weight of LaMnO<sub>3</sub> decreases (the percentage by weight of Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub> decreases), the crystal density of the Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub> phase also decreases, although it is relatively not large. The composition ratio by weight of Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub>-LaMnO<sub>3</sub> formed was 86%-14%, 74%-26%, and 63%-37%, respectively.



Figure 1. (a) X-ray diffraction pattern for samples YEG-1, YEG-2, and YEG-3. (b)-(d) Refine results using the Rietvelt program for samples YEG-1, YEG-2, and YEG-3, respectively.

The results of the analysis of the three samples have met the criteria of the Rietveld method with a value of  $\chi^2$  close to 1. This means that the sintering process of the mixture does not affect the phase change of the two composite-forming materials. However, it is hoped that with this sintering process, there will be intergrain interactions between Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub> and LaMnO<sub>3</sub> materials. This condition is very necessary for microwave absorbing materials because the microwave absorption mechanism is strongly influenced by the magnetic and electrical properties of the material [5]. Table 1 shows that the volume of the YEG-1, YEG-2, and YEG-3 composite V unit cells for the BaFe<sub>12</sub>O<sub>19</sub> phase ranges from 693.4 to 693.6 Å<sup>3</sup> with atomic density ranging from 5.874-5.876 g/cm3, while for the LaMnO<sub>3</sub> phase it ranges from 240.7 to 240.9 Å<sup>3</sup>. with atomic densities ranging from 6.666 to 6.673 g/cm<sup>3</sup>. This means that these two materials do not undergo structural changes after being composited through the sintering process.

Table 1. Resume of dif	fraction pattern and	dvsis results usin	g the Rietvel	t program.

	Phase	System	SG	Lattice parameters (Å)			V	ρ		fraction
Sample				а	b	с	(Å <sup>3</sup> )	(g/cm <sup>3</sup> )	X²	%
YEG-1	BaFe <sub>12</sub> O <sub>19</sub>	Hexagonal	P63/mmc	5.8851(3)	5.8851(3)	23.118(2)	693.4(1)	5.876		85.93
	LaMnO <sub>3</sub>	Orthorombic	Pbnm	5.545(5)	5.555(6)	7.815(5)	240.7(3)	6.673	1.036	14.07
YEG-2	BaFe <sub>12</sub> O <sub>19</sub>	Hexagonal	P63/mmc	5.8852(3)	5.8852(3)	23.122(2)	693.5(1)	5.875		73.86
	LaMnO <sub>3</sub>	Orthorombic	Pbnm	5.550(2)	5.531(3)	7.847(4)	240.9(1)	6.666	1.036	26.14
YEG-3	BaFe <sub>12</sub> O <sub>19</sub>	Hexagonal	P63/mmc	5.8853(2)	5.8853(2)	23.123(2)	693.6(1)	5.874		63.31
	LaMnO <sub>3</sub>	Orthorombic	Pbnm	5.551(6)	5.545(7)	7.825(2)	240.9(1)	6.667	1.032	36.69

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The results of the XRD analysis are also supported based on observations of the microstructure of this composite which is expected to produce a homogeneous mixture and have a relatively uniform particle size as shown in Figure 2. Figure 2 (a)-(c) shows a scanning electron microscopy (SEM) image of the microstructure of the  $Ba_{0.6}Sr_{0.4}Fe_{12}O_{19}$  composite with perovskite LaMnO<sub>3</sub> with various weight percentage fractions of 85/15, 75/25, and 65/35. The granules form like platelets, as seen in the insert in Figure 2 (c). The grain size ranges from 200 to 450 nm. This measure is obtained by comparing the grain length to the existing scale.  $Ba_{0.6}Sr_{0.4}Fe_{12}O_{19}$  phase and LaMnO<sub>3</sub> phase were mixed evenly. However, we suspect that based on the crystal size results, the small grains represent the LaMnO<sub>3</sub> phase, while the larger grains represent the  $Ba_{0.6}Sr_{0.4}Fe_{12}O_{19}$  phase. Similar results were also obtained by J. N. Dahal et al., where the hexaferrite particle size was larger than the perovskite particle size [7]. The size of the particles formed is also influenced by the sintering temperature. The higher the temperature, the larger the particle size that's formed [16].



(a) YEG-1



(b) YEG-2



## (c) YEG-3

Figure 2. Surface morphology of SEM results for samples (a) YEG-1, (b) YEG-2, and (c) YEG-3.

From the observation of the microstructure, it is shown that these three composites are homogeneous mixtures with relatively uniform particle sizes. This means that the distribution of particles in the mixture between  $BaFe_{12}O_{19}$  and  $LaMnO_3$  is evenly distributed over the entire sample surface in these three composites. Besides that, it also appears that the particles are not dispersed but there is good interconnection between the particles so that intergrain interactions are expected between the  $BaFe_{12}O_{19}$  and  $LaMnO_3$  materials. The presence of this intergrain interaction will greatly affect the magnetic properties and microwave absorption characteristics of the composite, especially in composites consisting of a mixture of materials that have much different characteristics [17].  $BaFe_{12}O_{19}$  material is a ferrous magnetic material that has hard magnetic characteristics with relatively large magnetic properties so that it tends to have a relatively large permeability value, while LaMnO<sub>3</sub> material is an antiferrous magnetic material but has a relatively high permittivity value.





Figure 3. Magnetic hysteresis loop curves of YEG-1, YEG-2, and YEG-3 samples at room temperature.

Magnetic hysteresis loop curves at room temperature for YEG-1, YEG-2 and YEG-3 composites are shown in Figure 3. The three samples have remanent magnetic magnitude Mr and coercivity field Hc are relatively the same, but differ in magnetic saturation values Ms. Compared to the other two samples, YEG-2 sample has the highest Ms saturation magnetic value, around 46.83 emu/g. The complete results of VSM can be seen in Table 2. At room temperature  $Ba_{0.6}Sr_{0.4}Fe_{12}O_{19}$  is hard magnetic, has a magnetic saturation of Ms of 92.9 emu/g and a coercive field of Hc = 4.24 kOe [4, 14], and LaMnO<sub>3</sub> has a value of Ms ~ 3.42 emu/g and Hc ~ 0.755 kOe are soft magnetic [7, 18, 19]. The  $Ba_{0.6}Sr_{0.4}Fe_{12}O_{19}$ -LaMnO<sub>3</sub> composite is a hard-soft magnetic. The magnitude of the magnetization can be seen in Table 2.

Table 2. Magnetization of Ms, Mr and Hc at room temperature					
Sample	Ms (emu/g)	Mr (emu/g)	Hc (kOe)		
YEG-1	46.17	28.80	3.88		

28.80

28.80

3.88

3.88

46.83

46.83

YEG-2

YEG-3

Based on the measurement results in Table 2, it appears that the magnetic properties of this material are relatively unchanged, this indicates that the level of particle homogeneity and the effect of the presence of intergrain interactions in the particles plays an important role in maintaining the magnetic properties of this composite. It is also suspected that this LaMnO<sub>3</sub> system when interacting with  $Ba^{2+}$  ions will be ferromagneticly so that it is able to maintain the Ms of this composite even though the  $BaFe_{12}O_{19}$  fraction is decreasing [20]. It is different if this composite is purely mixed without any sintering process, in general the saturation magnetization of Ms tends to decrease due to the reduced mass fraction of the  $BaFe_{12}O_{19}$  phase in the composite. These results are in agreement with the results of the previous analysis using XRD and microstructure observations using SEM. This study can also be seen from the measurement results of microwave absorption, as the main goal of this research.

The ability of a material to absorb electromagnetic waves is usually referred to as the reflection loss of electromagnetic radiation, RL (dB), calculated by the following equation: [21-22]

 $RL(dB)=20 \log |(Z_{in} - Z_0)/(Z_{in} + Z_0)|$ (3) where  $Z_{in}$  and  $Z_0$  represent the characteristic impedance of the material and the characteristic impedance of vacuum, respectively. The Zin value is obtained from the equation:

 $Z_{in} = (\mu_r / \varepsilon_r)^{(1/2)} \tanh[j(2\pi f d/c) (\mu_r \varepsilon_r)^{(1/2)}]$ (4)

where  $\mu_r$  and  $\mathcal{E}_r$  are the relative permeability and permittivity of the material, d is the thickness of the sample, c is the speed of electromagnetic waves in a vacuum, and f is the frequency with which the waves arrive. According to equation (3), when RL = -10 dB, the wave absorption reaches 90%.



Figure 4. Frequency dependent reflection loss (RL) for YEG-1, YEG-2, and YEG-3 samples.

The reflection loss for the YEG-1, YEG-2, and YEG-3 samples can be seen in Figure 4. The best minimum reflection loss is obtained for the YEG-2 sample, which is -13.0 dB, f = 11.12 GHz, and 1.52 GHz bandwidth. The increase in the weight percentage of LaMnO<sub>3</sub> in the Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub>-LaMnO<sub>3</sub> composite, although only slightly, the bandwidth increased from 1.42 GHz to 1.52 GHz, there was also a shift in the peak frequency from 10.96 GHz to 11.22 GHz. The amount of reflection loss (RL) by the absorbing material is influenced by electromagnetic parameters, such as permittivity and permeability of the material. Therefore, the soft magnetic properties of LaMnO<sub>3</sub> strengthen the microwave attenuation performance in the high frequency region [20]. While Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub> is a hard magnetic having a high Ms, so it will widen the absorption band [23].

Sample	RL max (dB)	f(GHz)	RL (dB)	Freq. range (GHz)
			-6.92	8.0-12.0 (4.0)
YEG-1	-11.21	8.12	-10.00	8.0-8.36 (0.36)
120-1	-10.62	8.96	-10.19	8.64-9.44 (0.8)
	-12.71	10.96	-11.62	10.0-11.42 (1.42)
			-8.0	8.0-12.0 (4.0)
YEG-2	-10.61	8.04	-10.30	8.0-8.14 (0.14)
	-13.91	9.32	-10.10	8.64-9.70 (1.06)
	-13.02	11.12	-10.04	9.98-11.50 (1.52)
			-6.81	8.0-12.0 (4.0)
YEG-3	-11.52	9.50	-10.11	9.04-9.68 (0.64)
	-11.61	11.22	-10.10	9.98-11.50 (1.52)

Table 3. Value of reflection loss (RL), frequency and bandwidth for YEG-1, YEG-2, dan YEG-3 samples

In Table 3, it can be seen that the performance improvement of this composite-based microwave absorbing material is obtained, namely reflection loss and the width of the microwave absorption area. It is known that before being modified these two materials have low reflection loss, but when they become composites and it is strongly suspected that interconnection occurs between grains, this results in the composite having higher permeability and permittivity so as to produce higher microwave absorption. The most

interesting thing is that the absorption frequency area becomes wider. The YEG-2 composite found microwave absorption with RL = -8 dB at a wide frequency from 8-12 GHz. This composition becomes a reference and initial study for the development of further composite materials, so that it is expected that composite materials with more optimum absorption capabilities can be obtained.

#### 4. CONCLUSION

The Ba<sub>0.6</sub>Sr<sub>0.4</sub>Fe<sub>12</sub>O<sub>19</sub>/LaMnO<sub>3</sub> composite material which has been successfully synthesized by mechanical alloying method has a hexagonal/orthorhombic crystal structure with a homogeneous mixture and relatively uniform particle size of about 200 to 450 nm. This sintering process produces interconnections between particles in the composite so that it is suspected. Intergrain interactions occur which are able to maintain their magnetic properties even though the BaFe<sub>12</sub>O<sub>19</sub> fraction is reduced. Besides that, it produces a significant increase in microwave absorption capability with a relatively wide absorption frequency area of 8-12 GHz with a reflection loss of -8 dB. This research can be used as a reference and initial study for the development of further composite materials in order to obtain composite materials with more optimum absorption capabilities.

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